

Nanometer Arrays of Functional Light Harvesting Antenna Complexes by Nanoimprint Lithography and Host#Guest Interactions

Maryana Escalante, Yiping Zhao, Manon J. W. Ludden, Rolf Vermeij, John D. Olsen, Erwin Berenschot, C. Neil Hunter, Jurriaan Huskens, Vinod Subramaniam, and Cees Otto

J. Am. Chem. Soc., **2008**, 130 (28), 8892-8893 • DOI: 10.1021/ja802843m • Publication Date (Web): 21 June 2008

Downloaded from <http://pubs.acs.org> on December 19, 2008

More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

[View the Full Text HTML](#)



Nanometer Arrays of Functional Light Harvesting Antenna Complexes by Nanoimprint Lithography and Host–Guest Interactions

Maryana Escalante,[†] Yiping Zhao,^{‡,§} Manon J. W. Ludden,[‡] Rolf Vermeij,[†] John D. Olsen,^{||} Erwin Berenschot,[§] C. Neil Hunter,^{||} Jurriaan Huskens,^{*,‡} Vinod Subramaniam,^{*,†} and Cees Otto^{*,†}

Biophysical Engineering Group, Molecular Nanofabrication Group, Transducer Science and Technology Group, and MESA+ Institute for Nanotechnology, University of Twente P.O. Box 217, 7500 AE Enschede, The Netherlands, and Department of Molecular Biology and Biotechnology, University of Sheffield, Sheffield S10 2TN, U.K.

Received April 17, 2008; E-mail: c.otto@utwente.nl; v.subramaniam@utwente.nl; j.huskens@utwente.nl

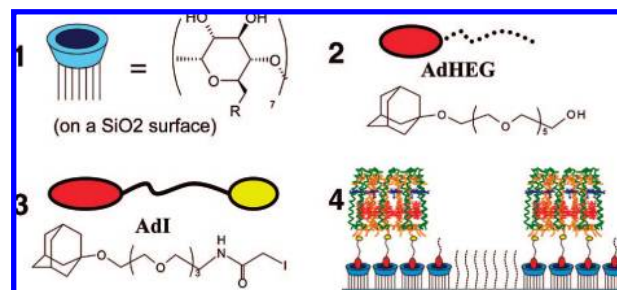
Supramolecular interactions play a key role in the functional architecture of nature. On patterned surfaces, interactions can be adjusted in strength and further modulated by the orientation of target molecules.^{1,2} Here, we have engineered functional ordered arrays of purified components of the photosynthetic system. We relied on multivalent interactions to drive the selective assembly of functional light harvesting LH2 antenna complex onto nanometer structured β -cyclodextrin (β -CD) monolayers² patterned by nanoimprint lithography (NIL).^{3,4}

The nanomachinery of the photosynthetic bacterium *Rhodospira rubra* has been an invaluable model for the study of biophysics, biochemistry, and molecular biology of photosynthesis.⁵ The membrane-bound LH2 complex is built of nine identical subunits each consisting of an α and a β polypeptide. A total of 27 bacteriochlorophyll (BChl) molecules (18 BChl B850 and 9 B800) are bound to this structure having dimensions of ~ 6 nm in height and ~ 6 nm in diameter.^{6,7} LH2 are interesting candidates for applications in synthetic light converting circuits because of their well-defined optical properties, such as a broad spectral range, high absorption cross section, efficient energy transfer⁸ and high photostability. Photosynthetic antenna systems have been used in studies exploiting covalent⁹ and electrostatic interactions¹⁰ to promote attachment to a chemically defined surface. A major challenge remains in the control of the interfacial properties and the associated multiple weak interactions to produce and optimize organized molecular structures with controlled directional energy migration.

Here, we show an approach based on a combination of site-directed mutagenesis, NIL and multivalent host–guest interactions. LH2 complexes were engineered with cysteine residues at the penultimate position of the C-terminus of the α polypeptide chain. These strategic positions at the periplasmic face ensured the orientation of all of the protein complexes upon binding to the surface. The cysteine residues were modified with iodoacetyl-tri(ethylene glycol) mono(adamantyl ether), AdI, block 3, Chart 1. Protein aggregates in an aqueous buffered solution of 20 mM HEPES, pH 8.0, 0.03% *n*-dodecyl- β -D-maltoside (β -DDM) were mixed in 1:20 mol equiv with the AdI in 1.3% dimethyl sulfoxide (DSMO) to yield an average of three adamantyl molecules linked to each protein complex, hereafter referred to as Ad_nLH2.¹¹

When adsorbing Ad_nLH2 onto a β CD-coated glass substrate (Chart 1, block 1), hexa(ethylene glycol)mono(adamantyl ether) AdHEG, block 2, served as a temporary blocking agent for the

Chart 1. Representation of Host, Guest, and Target Molecules^a



^a (1) β -CD heptamine, host molecule,² (2) Hexa(ethylene glycol)-mono(adamantyl ether) (AdHEG), (3) iodoacetyl-tri(ethylene glycol) mono(adamantyl ether), (AdI), (4) Ad_nLH2 on the β -CD monolayer.

β -CD cavities, preventing nonspecific adsorption by shielding the surface with the HEG chain.¹² The monovalent AdHEG is later effectively displaced through competition by exploiting the higher affinity of the multivalent Ad_nLH2.¹² The assembly on the patterned surface is depicted in block 4.

We used a custom-built hybrid high resolution scanning probe-spectral microscope¹³ to characterize the patterned proteins in liquid conditions. A fluorescence titration allowed us to simultaneously address the optical properties after modification of the LH2 complex and the specificity of the binding on nonpatterned β -CD surfaces. Upon excitation of the LH2 complexes via the B800 BChl (donor), the energy is then transferred within the complex to the B850 (acceptor) and ultimately emitted as fluorescence. A dilute solution of nonmodified LH2 complexes was incubated onto the β -CD monolayer and rinsed with buffer. The average emission spectra (blue box trace, Figure 1a), indicates a high contribution of nonspecific adsorption. Subsequently, after pretreating the surface with 1 mM solution of AdHEG, a solution of nonmodified LH2 complexes in 1 mM AdHEG was incubated onto the substrate. The nonspecific adsorption was reduced by 94% (open box trace). The previous experiment was repeated with Ad_nLH2 complexes. The increase in intensity of the emission signal (green star trace) reveals that the protein complexes specifically bind to the surface by replacement of the monovalent AdHEG with the multivalent Ad_nLH2. This replacement¹² and the observed stability against rinsing with a β CD solution² are evidence for the formation of stable multivalent complexes via at least 3 Ad linkers. Quantitatively, the spectral response from the immobilized Ad_nLH2 complexes showed no visible shift of the emission maximum, ~ 868 nm, with respect to bulk measurements of nonmodified LH2 complexes (red triangle trace). This observation is compelling evidence that the complete procedure of labeling and surface adsorption has maintained the

[†] Biophysical Engineering Group, University of Twente.

[‡] Molecular Nanofabrication Group, University of Twente.

[§] Transducer Science and Technology Group, University of Twente.

^{||} University of Sheffield.

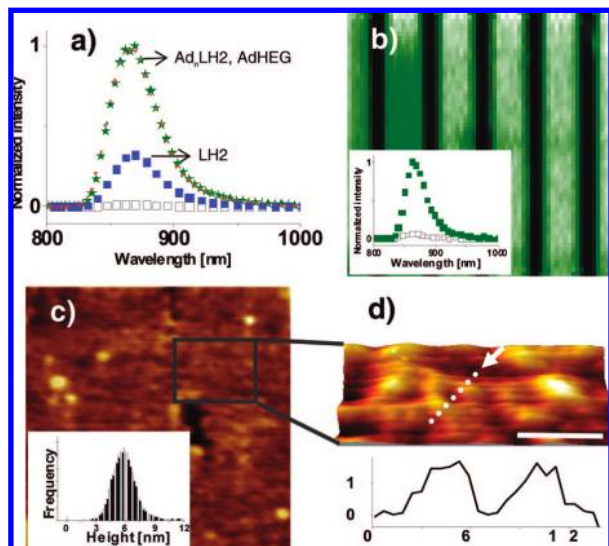


Figure 1. (a) Fluorescence titration: nonlabeled LH2 (blue box); nonlabeled LH2, 1 mM AdHEG (open box); Ad_nLH2, 1 mM AdHEG (green star); reference spectrum of nonlabeled LH2 in solution (red triangle). (b) False color, fluorescent-spectral image of Ad_nLH2 patterns (β -CD/PEG), $40 \times 40 \mu\text{m}$, 64×64 pixels; inset shows emission spectra active area (green box), passivated (open box). (c) AFM topography in liquid, $150 \times 150 \text{ nm}$ area, 256×256 pixels, inset shows histogram height distribution. (d) Section across a LH2 complex showing a profile along the dotted line; scale bar, 10 nm.

structural integrity of the LH2 membrane protein. Figure 1b shows a representative spectral image for the patterned Ad_nLH2 complexes. On each pixel, a full spectrum was recorded, and integrated over the respective emission band of the LH2 complexes. The exposure of patterns of β -CD SAMs surrounded by a protein resistant PEG SAM¹⁴ to the solution of Ad_nLH2, 1 mM AdHEG resulted in the selective assembly of the protein onto the β -CD regions (green) in a ratio of 16:1 as indicated by the averaged emission spectra (inset Figure 1b) from active and passivated areas. A monolayer coverage was suggested by quantitative spectral images, which revealed only minor variations ($<5\%$) in intensity over the patterned area.

We performed AFM imaging at low tapping amplitudes to assess the density of the putative monolayer. Figure 1c shows an AFM height image of the patterned LH2 complexes; analysis of the surface indicates a uniform height of $\sim 6 \text{ nm}$ (inset) with reference to the defect (black region), suggesting monolayer coverage of the protein. In the AFM image (Figure 1d) ring-shaped LH2 proteins could be observed, which are attributed to the exposed face of the complexes. The height histogram (inset Figure 1c) indicated less than 1% multilayer stacked aggregates of proteins in agreement with the fluorescence images.

In an attempt to prepare structures approaching molecular dimensions, NIL was performed using stamps with silicon ridges as small as 40 nm, $4 \mu\text{m}$ period. Figure 2a shows an AFM height image in liquid of Ad_nLH2 complexes on a β -CD SAM. The mean fwhm of the lines after processing is $80 \pm 5 \text{ nm}$ with a height of $\sim 6 \pm 1 \text{ nm}$ consistent with the assembly of a monolayer of LH2 complexes, Figure 2c. The increase in width, relative to that of the NIL stamp, is attributed to the process of removal of the residual layer in the imprinting process. Figure 2b shows a fluorescence image acquired with a single photon counting avalanche photodiode, with intensity variations of $\pm 14\%$ along the lines. The apparent

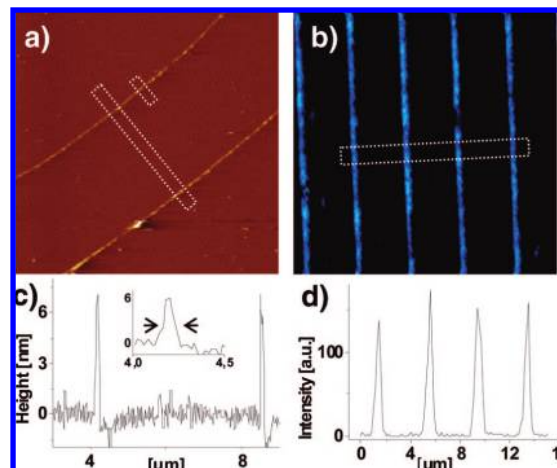


Figure 2. (a) AFM topography in liquid of Ad_nLH2 β -CD/PEG SAM, $10 \times 10 \mu\text{m}$, z-scale 30 nm, and respective cross section (c), fwhm of 80 nm (inset). (b) False color fluorescent image. (d) Cross section of panel b. width of these structures (Figure 2d) is defined by the optical resolution of $\sim 700 \text{ nm}$.

We have achieved exquisite spatial control at different length scales of functional specifically bound LH2 complexes in a high throughput manner by exploiting host–guest interactions and NIL. In situ characterization of the formation of these assemblies at molecular dimensions and the fabrication of mixed protein arrays are the subject of current research.

Acknowledgment. This work was supported by NanoNed TMM.7124; NWO-CW (M.J.W.L.; Vidi Vernieuwingsimpuls 700.52.423 to J.H.); MESA+ (SRO-NF) (Y.Z.); C.N.H. and J.D.O. acknowledge funding from the BBSRC, U.K.

Supporting Information Available: Materials and experimental procedures. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) Mulder, A.; Onclin, S.; Peter, M.; Hoogenboom, J. P.; Beijleveld, H.; ter Maat, J.; Garcia-Parajo, M. F.; Ravoo, B. J.; Huskens, J.; van Hulst, N. F.; Reinhoudt, D. N. *Small* **2005**, *1*, 242–253.
- (2) Ludden, M. J. W.; Reinhoudt, D. N.; Huskens, J. *J. Chem. Soc. Rev.* **2006**, *35*, 1122–1134.
- (3) Chou, S. Y.; Krauss, P. R.; Renstrom, P. J. *Science* **1996**, *272*, 85–87.
- (4) Maury, P.; Peter, M.; Crespo-Biel, O.; Ling, X. Y.; Reinhoudt, D. N.; Huskens, J. *Nanotechnology* **2007**, *18*, 044007.
- (5) Hunter, C. N.; Tucker, J. D.; Niederman, R. A. *Photochem. Photobiol.* **2005**, *4*, 1023–1027.
- (6) Scheuring, S.; Seguin, J.; Marco, S.; Levy, D.; Breytton, C.; Robert, B.; Rigaud, J. L. *J. Mol. Biol.* **2003**, *325*, 569–580.
- (7) Bahatryova, S.; Frese, R. N.; van der Werf, K. O.; Otto, C.; Hunter, C. N.; Olsen, J. D. *J. Biol. Chem.* **2004**, *279*, 21327–21333.
- (8) (a) Sundstrom, V.; Pullerits, T.; van Grondelle, R. *J. Phys. Chem B* **1999**, *103*, 2327–2346. (b) van Grondelle, R.; Novoderezhkin, V. I. *Phys. Chem. Chem. Phys.* **2006**, *8*, 793–807.
- (9) Reynolds, N. P.; Janusz, S.; Escalante-Marun, M.; Timney, J.; Ducker, R. E.; Olsen, J. D.; Otto, C.; Subramaniam, V.; Leggett, G. J.; Hunter, C. N. *J. Am. Chem. Soc.* **2007**, *129*, 14625–14631.
- (10) Escalante, M.; Maury, P.; Bruinink, C. M.; van der Werf, K.; Olsen, J. D.; Timney, J. A.; Huskens, J.; Hunter, C. N.; Subramaniam, V.; Otto, C. *Nanotechnology* **2008**, *19*, 025101.
- (11) Molecular Probes, Invitrogen Corporation, Carlsbad, CA.
- (12) Ludden, M. J. W.; Mulder, A.; Tampe, R.; Reinhoudt, D. N.; Huskens, J. *Angew. Chem., Int. Ed.* **2007**, *46*, 4104–4107.
- (13) Kassies, R.; Van der Werf, K. O.; Lenferink, A.; Hunter, C. N.; Olsen, J. D.; Subramaniam, V.; Otto, C. *J. Microsc. (Oxford)* **2005**, *217*, 109–116.
- (14) Kannan, B.; Castellino, K.; Chen, F.; Majumdar, A. *Biosens. Bioelectron.* **2006**, *21*, 1960–1967.

JA802843M